

# JS Organic Practical Course

Additional NMR spectra to aid structural determination

Three types of data are supplied :

- 2D homonuclear correlation spectra (HH COSY)
- 2D heteronuclear correlation spectra (CH COSY)
- 1D Nuclear Overhauser spectra (NOE)

Data is supplied for experiments :

- 2 (Benzoin), 3b (1,4-Diphenyl-1,3-butadiene)
- 5 (2,6-Dimethyl-3-heptan-5-one)
- 6a (1-Phenylbutane-1,4-diol)
- 6b ( $\gamma$ -Phenyl- $\gamma$ -butyrolactone)
- 7 (*cis*-Caran-*trans*-4-ol)

# Additional NMR analysis connections through bonds and space

- Correlation Spectra (COSY) – through bond connections
  - HH COSY - connections of proton spins through bonds
  - CH COSY- direct link of carbon to proton(s)
  - Long range CH COSY
- Connections through space
  - Nuclear Overhauser (NOE) experiments
    - 1D : Difference NOE, DPFGSE-NOE
    - 2D : NOESY, ROESY

# How to read a COSY

- HH COSY

diagonal is the 1D spectrum

off diagonal signal(s) display the connections of the spins

true signal *must* have mirror image across the diagonal

- CH COSY

signals are the direct correlation between the C and H

- Long range CH COSY

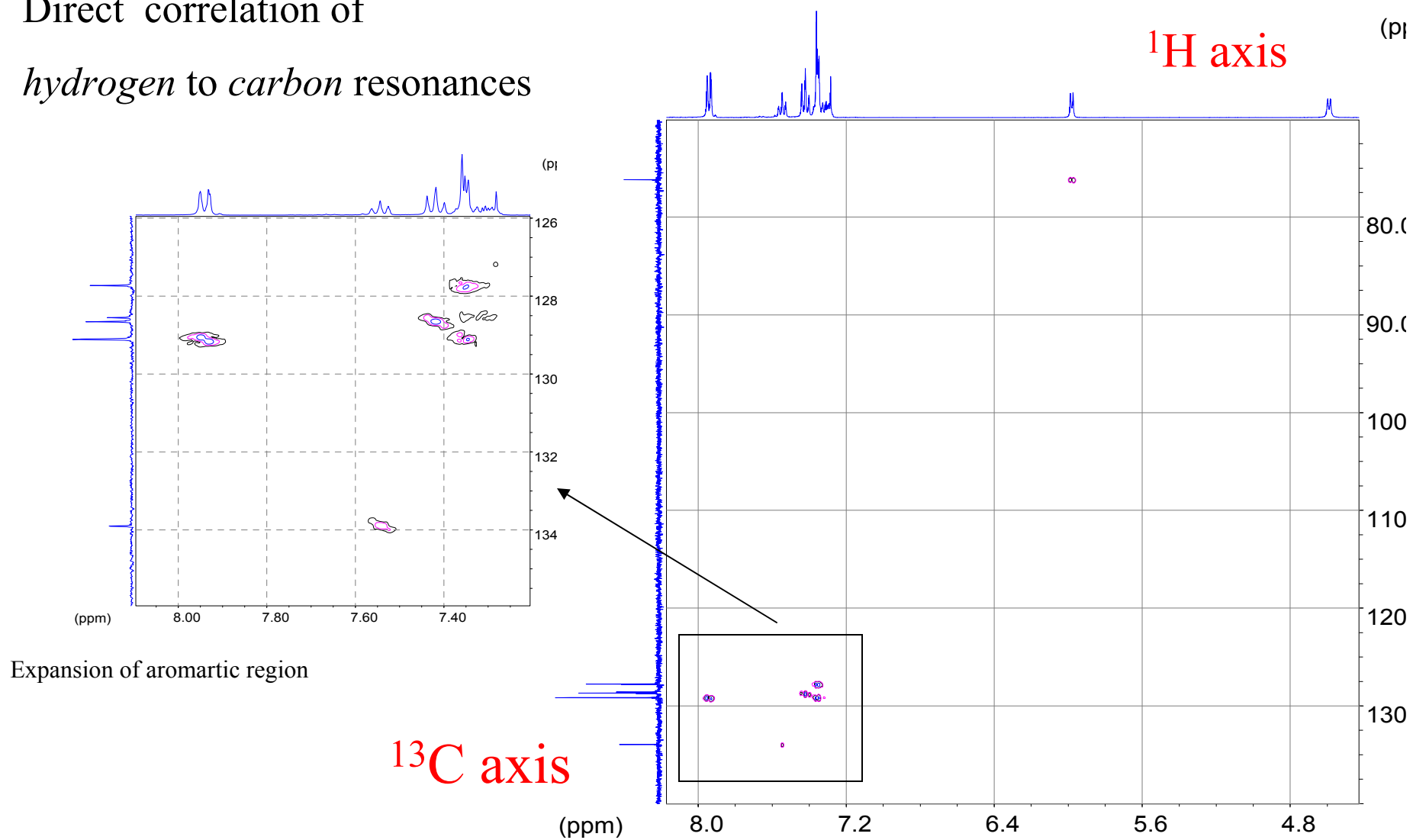
often can correlate several protons to a carbon

(or *vice versa* - whichever is most appropriate )

e.g. links carbon signals with NO protons directly attached

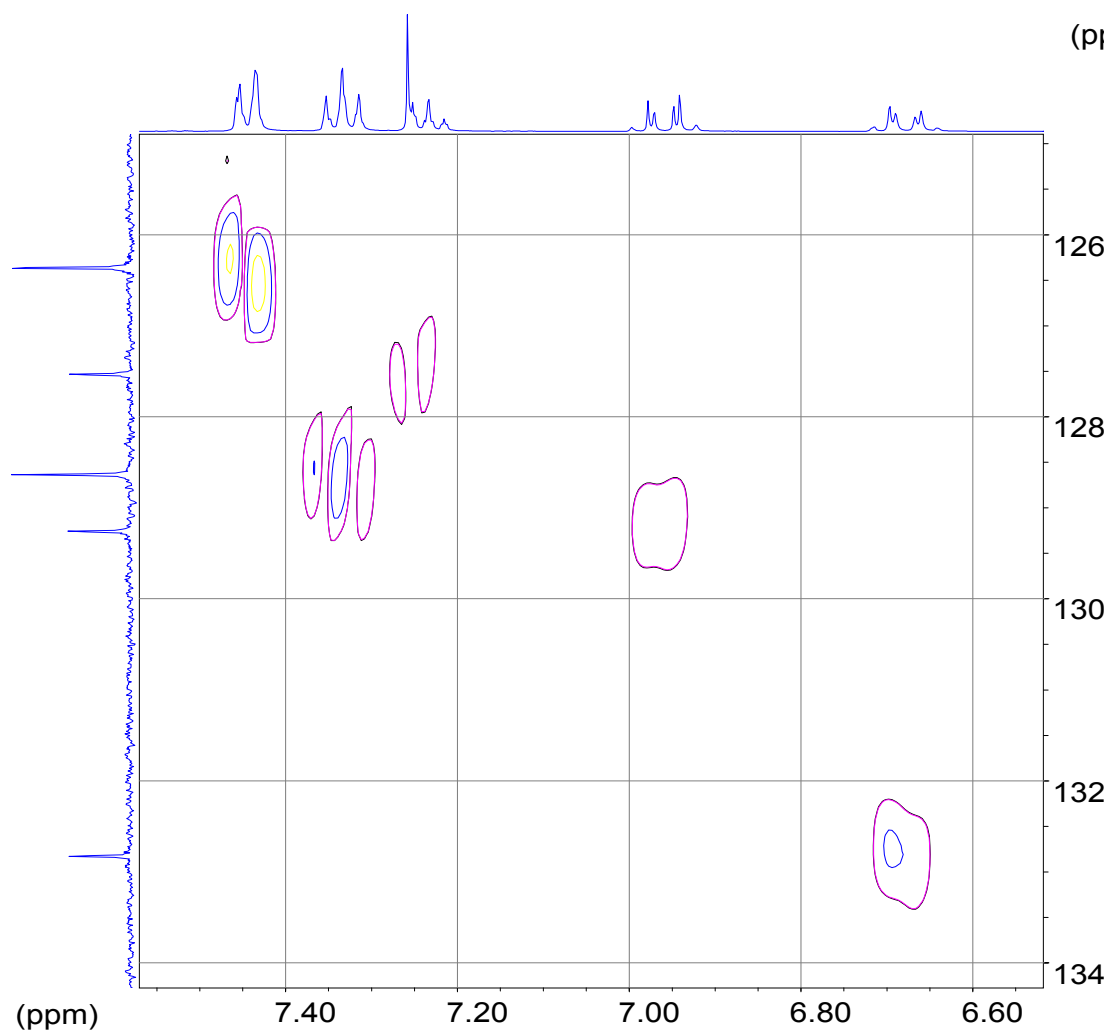
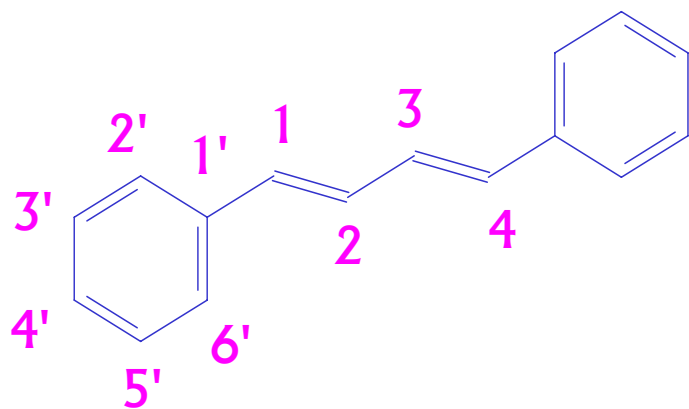
## 2 CH COSY Benzoin

Direct correlation of  
*hydrogen to carbon* resonances

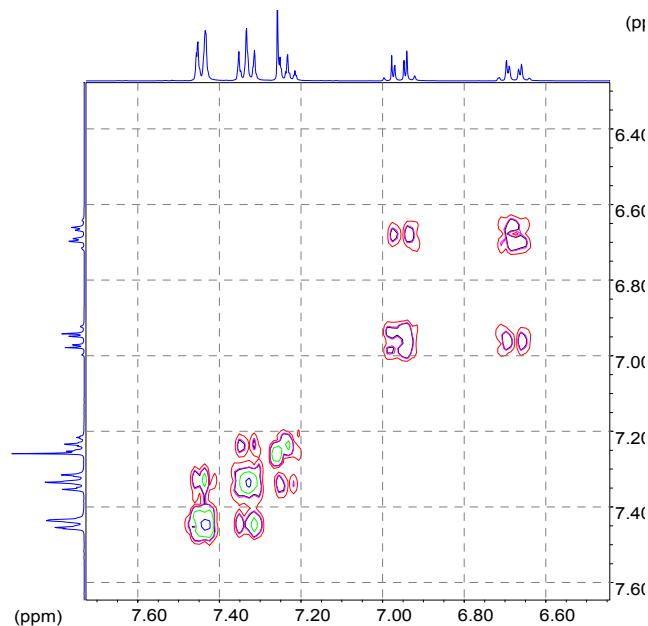


### 3B CH COSY (E,E)-1,4-Diphenyl-1,3-butadiene

Direct correlation of  
*hydrogen* to *carbon* resonances



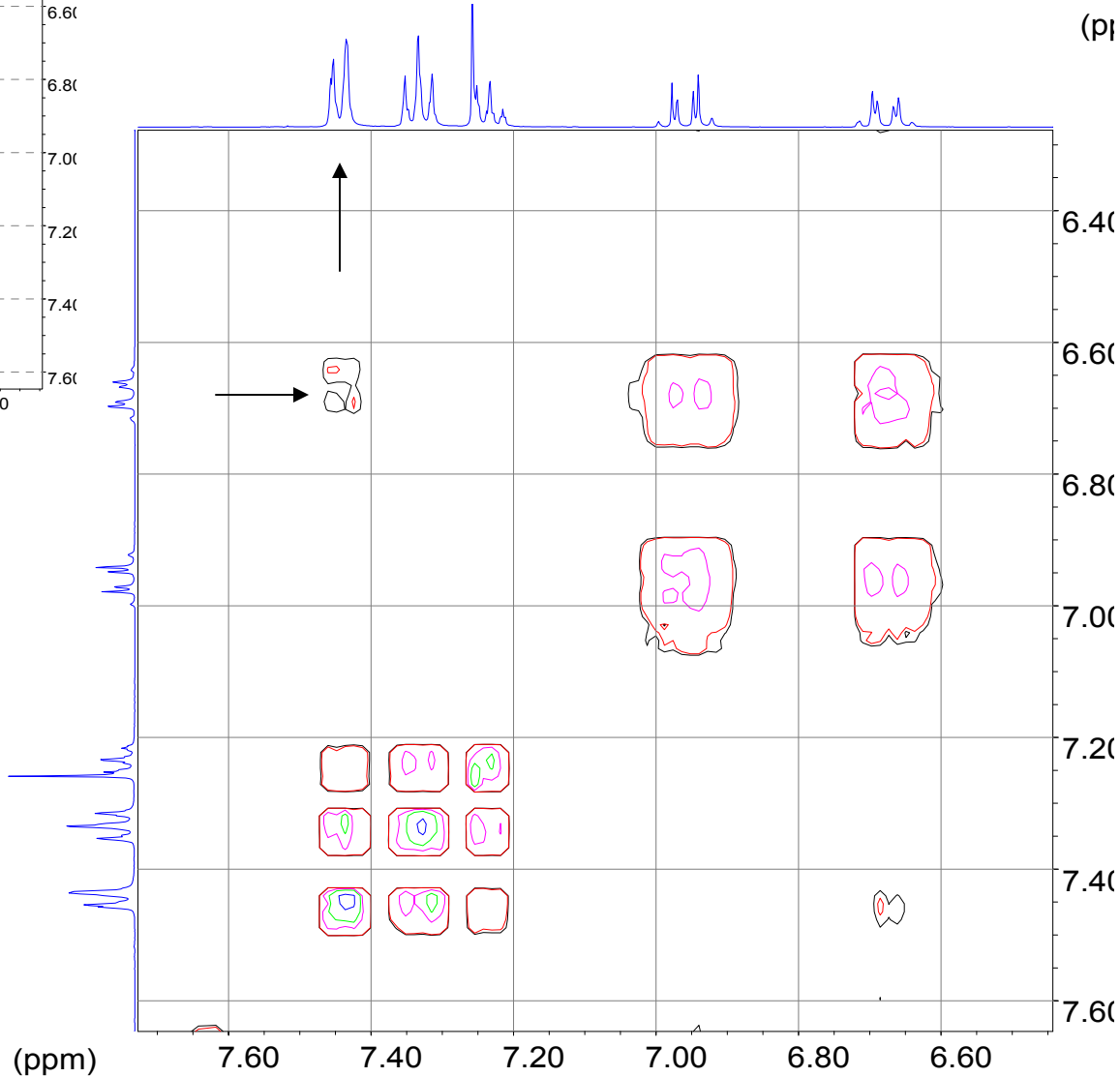
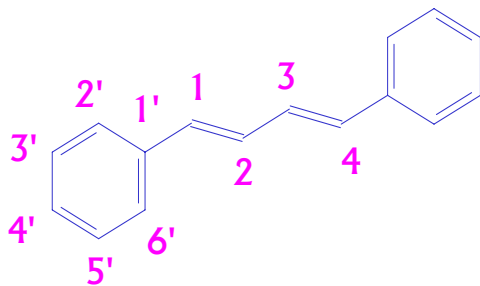
## 3B HH COSY (E,E)-1,4-Diphenyl-1,3-butadiene



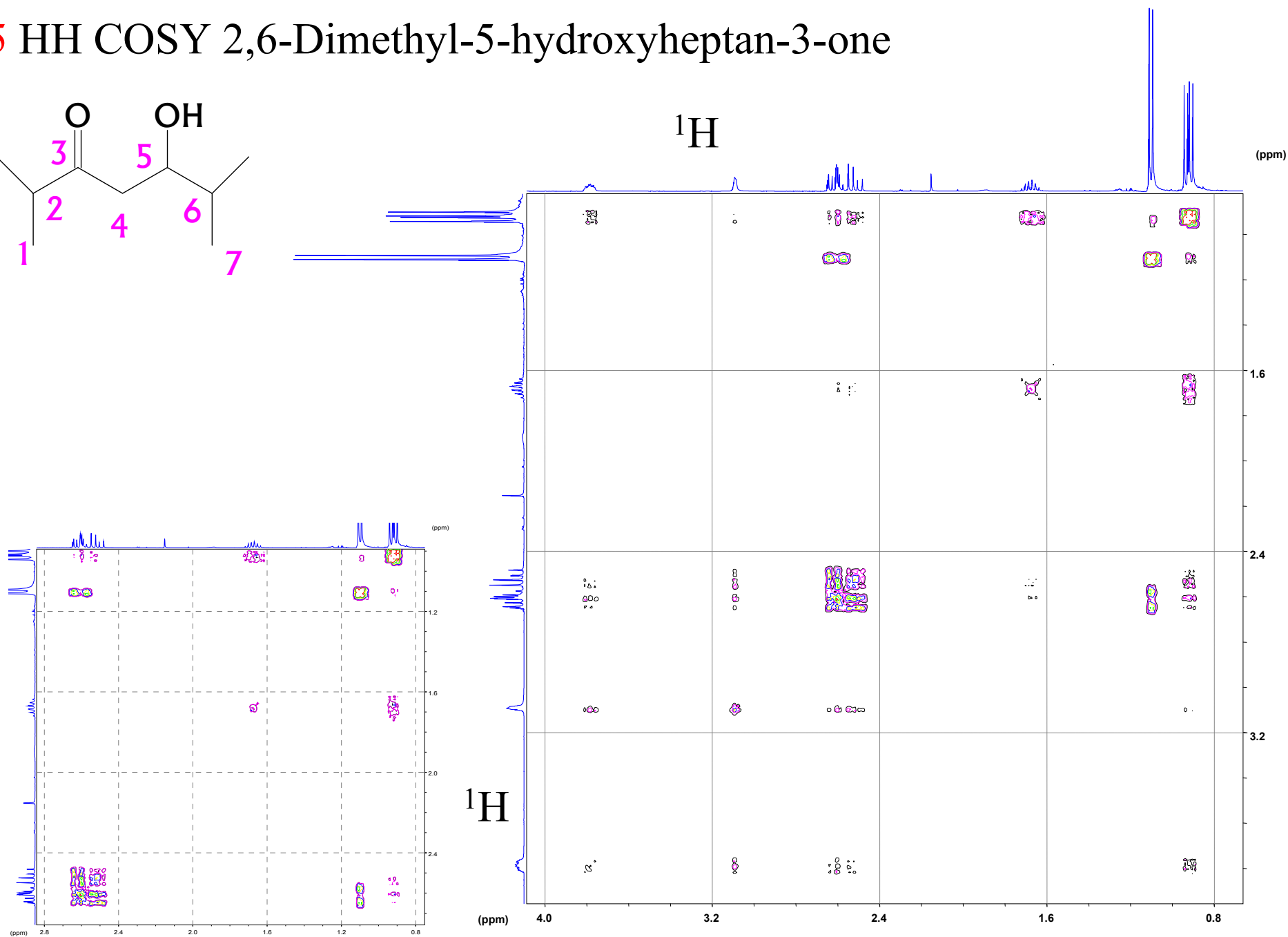
Diagonal contains the 1D spectrum

Connections between atoms are symmetrical about the diagonal

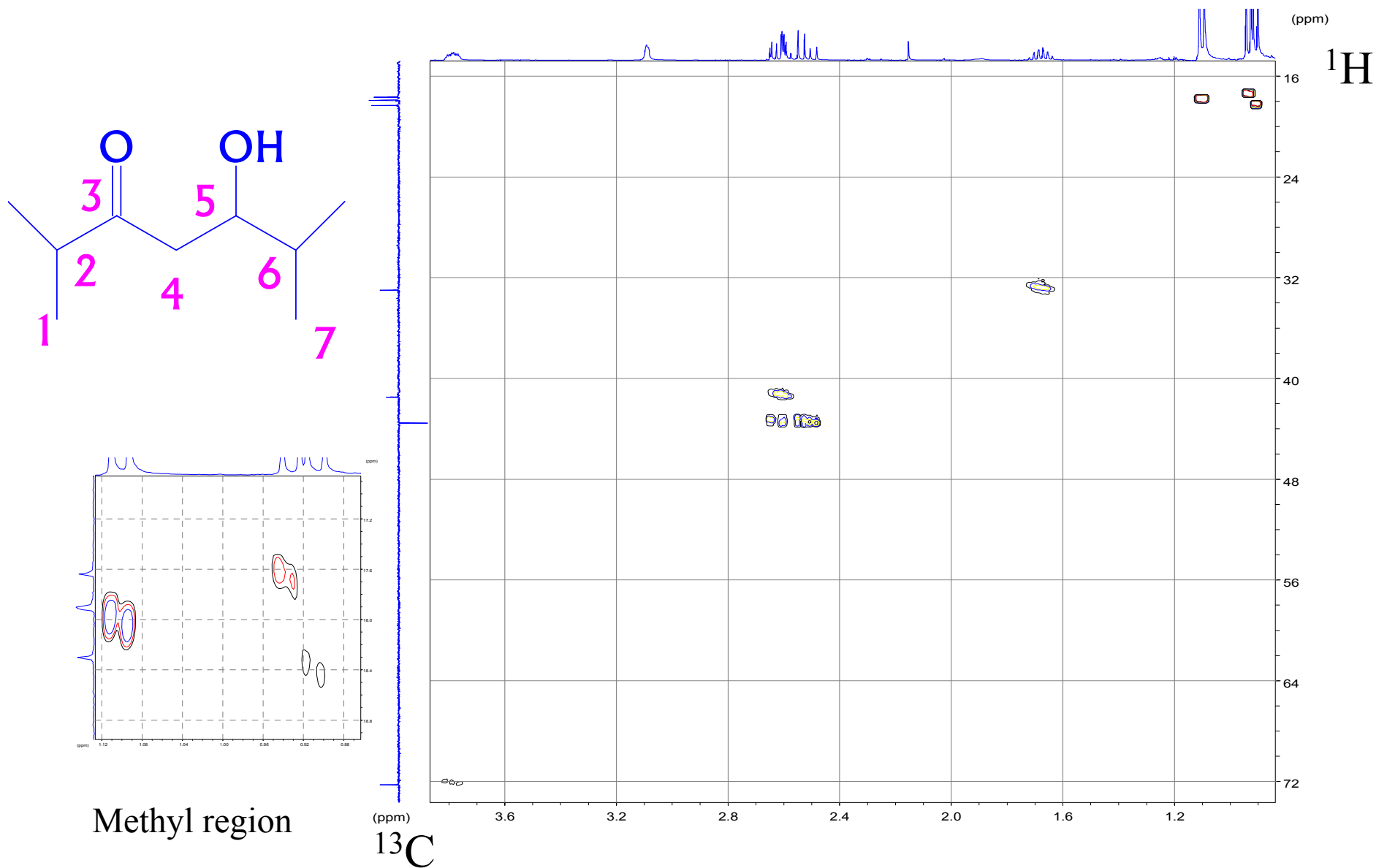
Generally, the higher the contour the stronger the connection



The chemical structure shows 3,5-dimethyl-4-hydroxy-2-pentanone. The carbons are numbered as follows: 1 is the methyl group on the left; 2 is the carbonyl carbon; 3 is the methyl group on the right of the carbonyl; 4 is the methine carbon bearing the hydroxyl group; 5 is the methyl group on the left of the hydroxyl-bearing carbon; 6 is the methyl group on the right of the hydroxyl-bearing carbon; and 7 is the terminal methyl group on the far right.



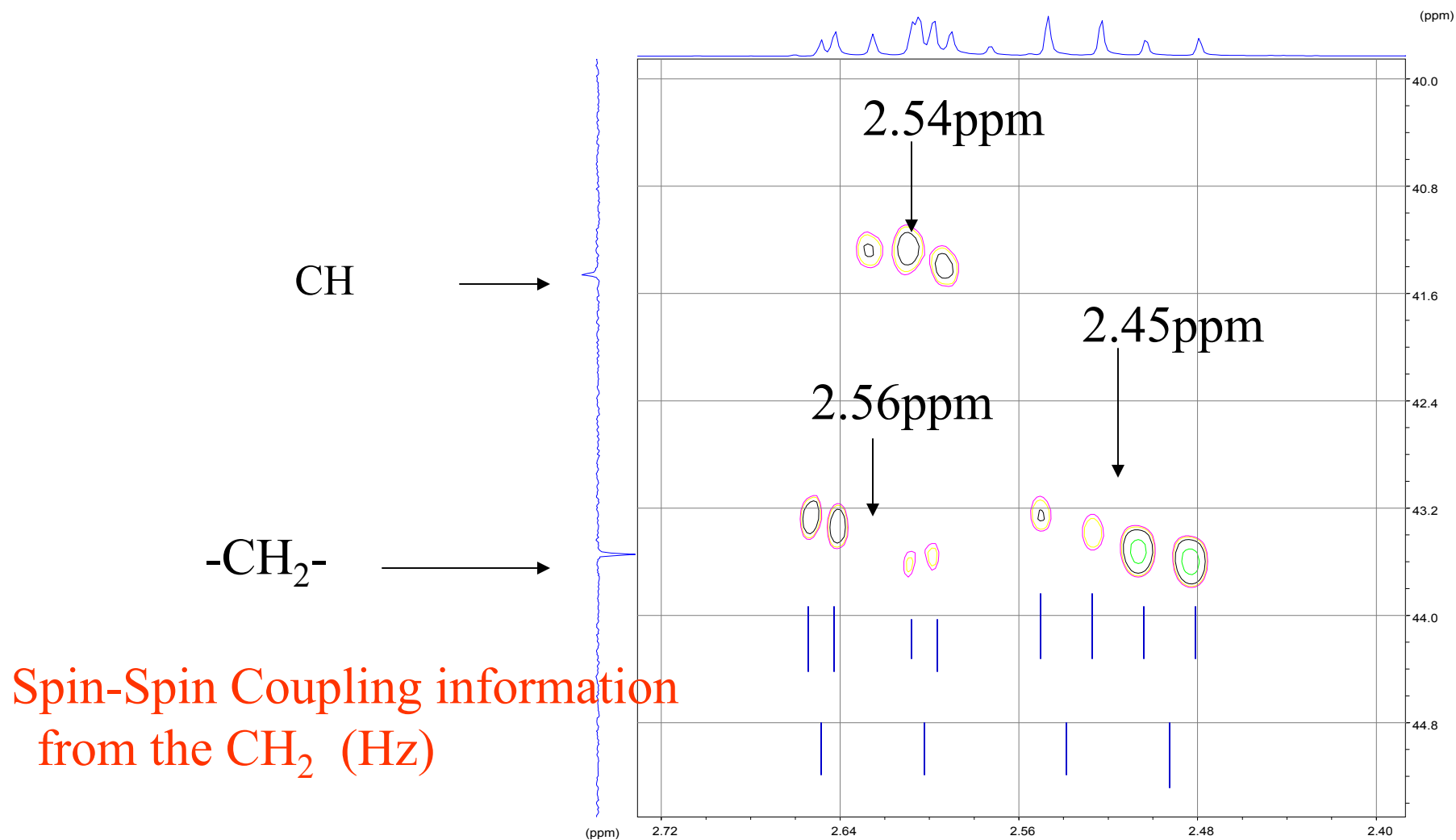
# 5 CH COSY 2,6-Dimethyl-5-hydroxyheptan-3-one





## 5 CH COSY 2,6-Dimethyl-5-hydroxy-heptan-3-one

Expansion of the 2.40-2.75ppm region

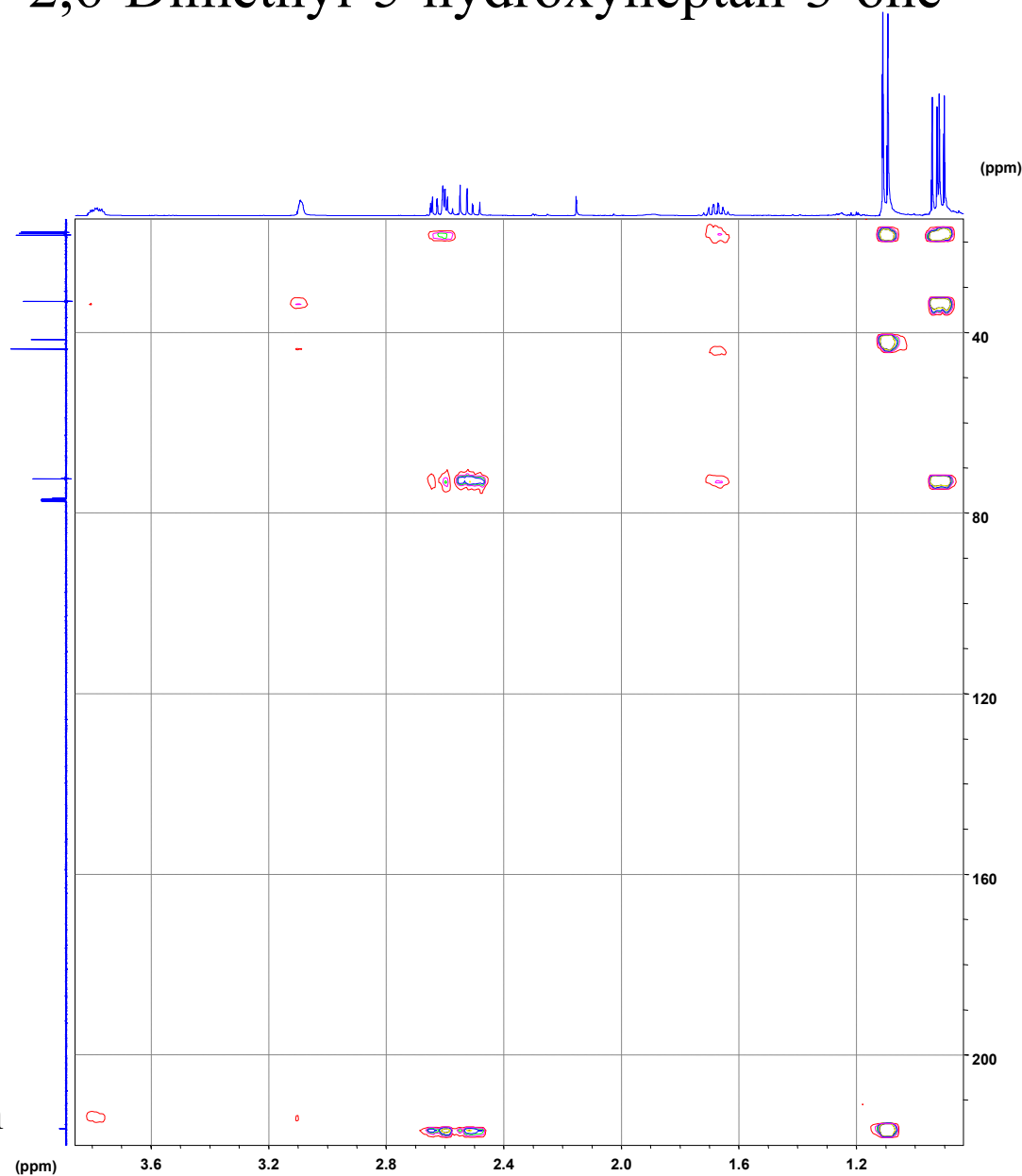


## 5 Long Range CH COSY 2,6-Dimethyl-5-hydroxyheptan-3-one

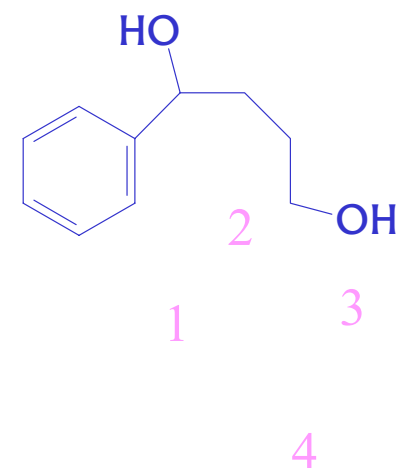
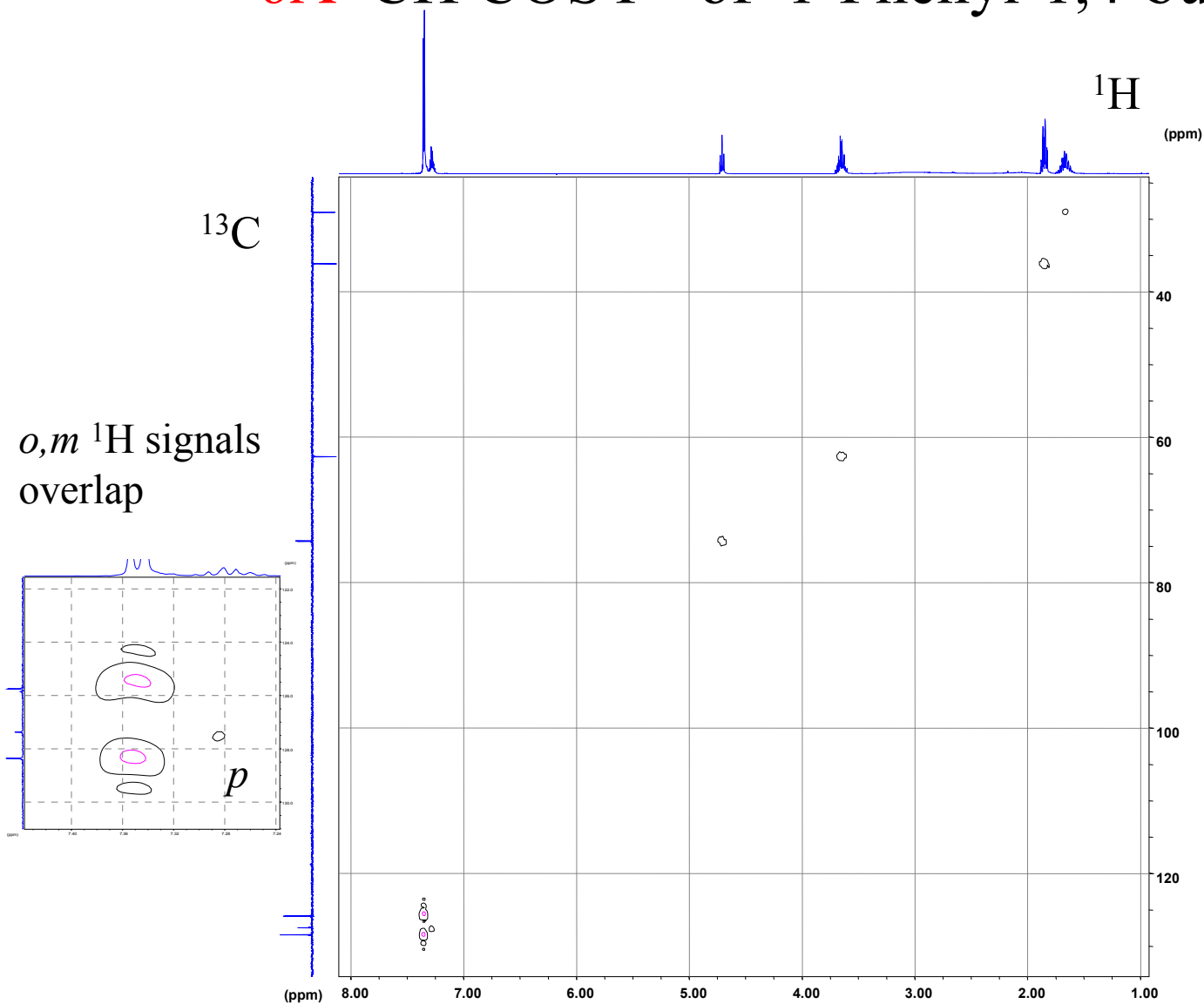
Notice CH correlations to :

- a longer range  
*i.e.* two or more bonds  
for a CH correlation
- a carbonyl resonance  
*i.e.* a quaternary peak
- a hydroxyl peak

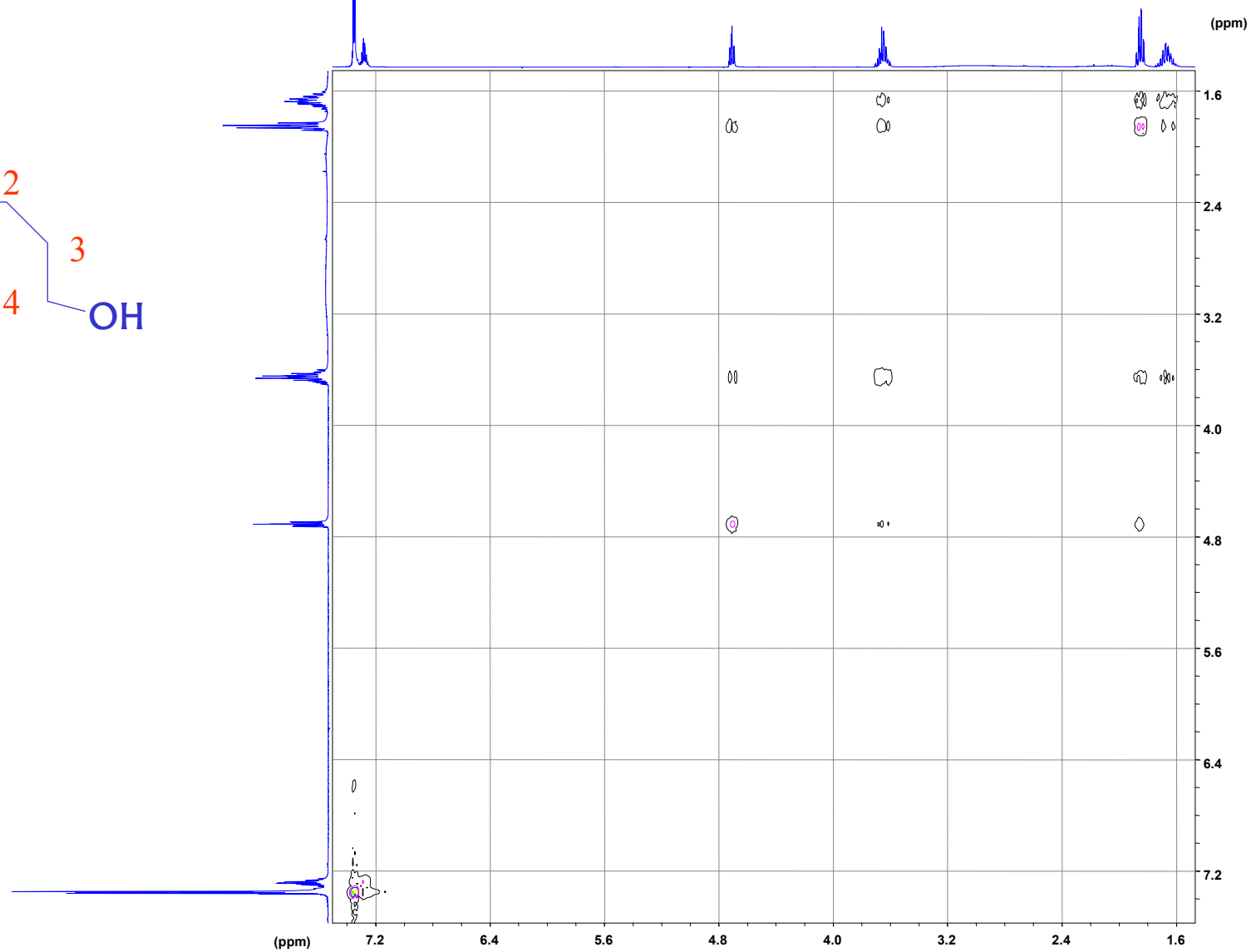
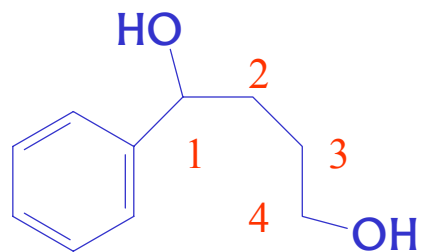
Keto Carbon 216 ppm



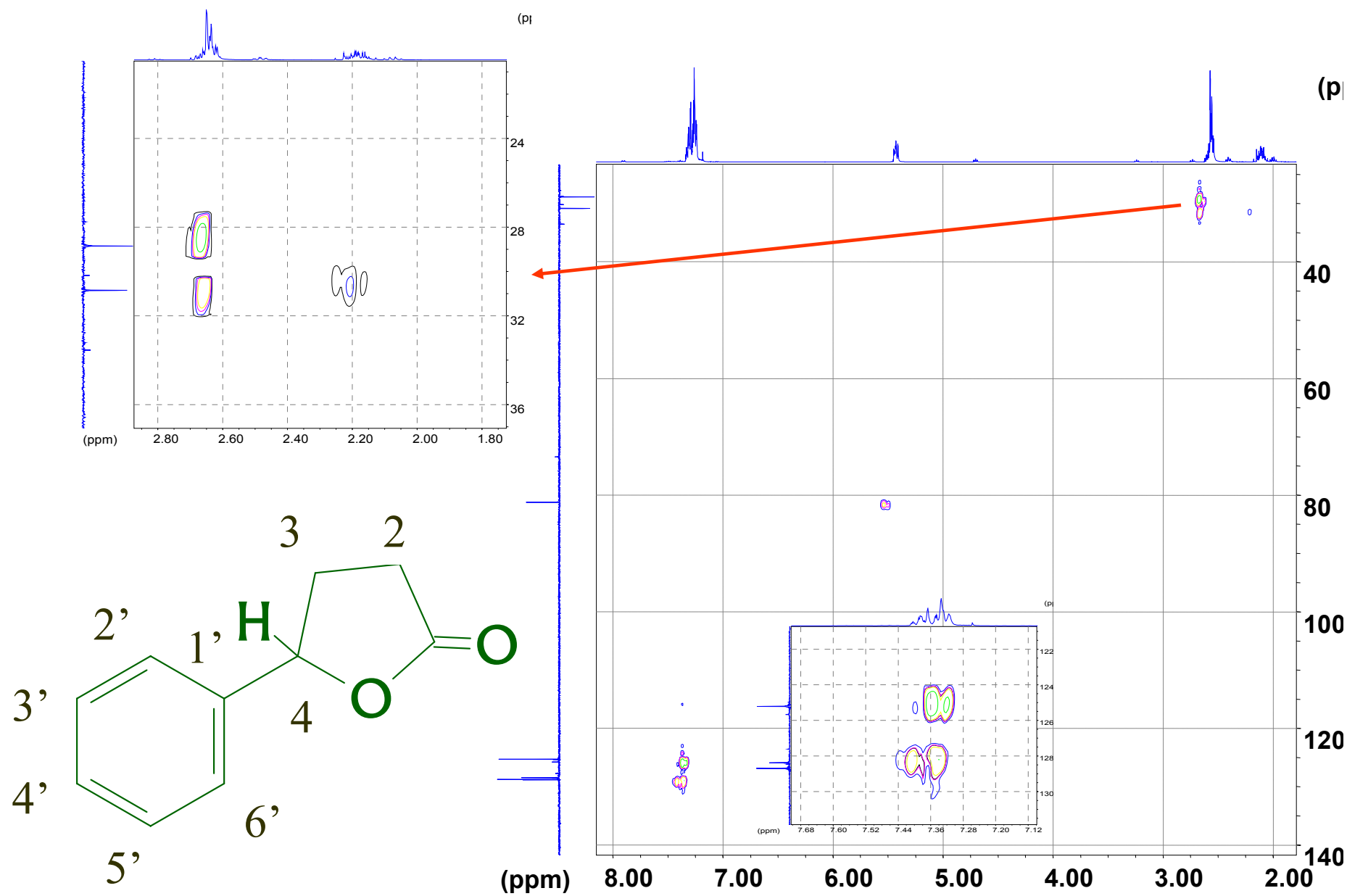
# 6A CH COSY of 1-Phenyl-1,4-butanediol



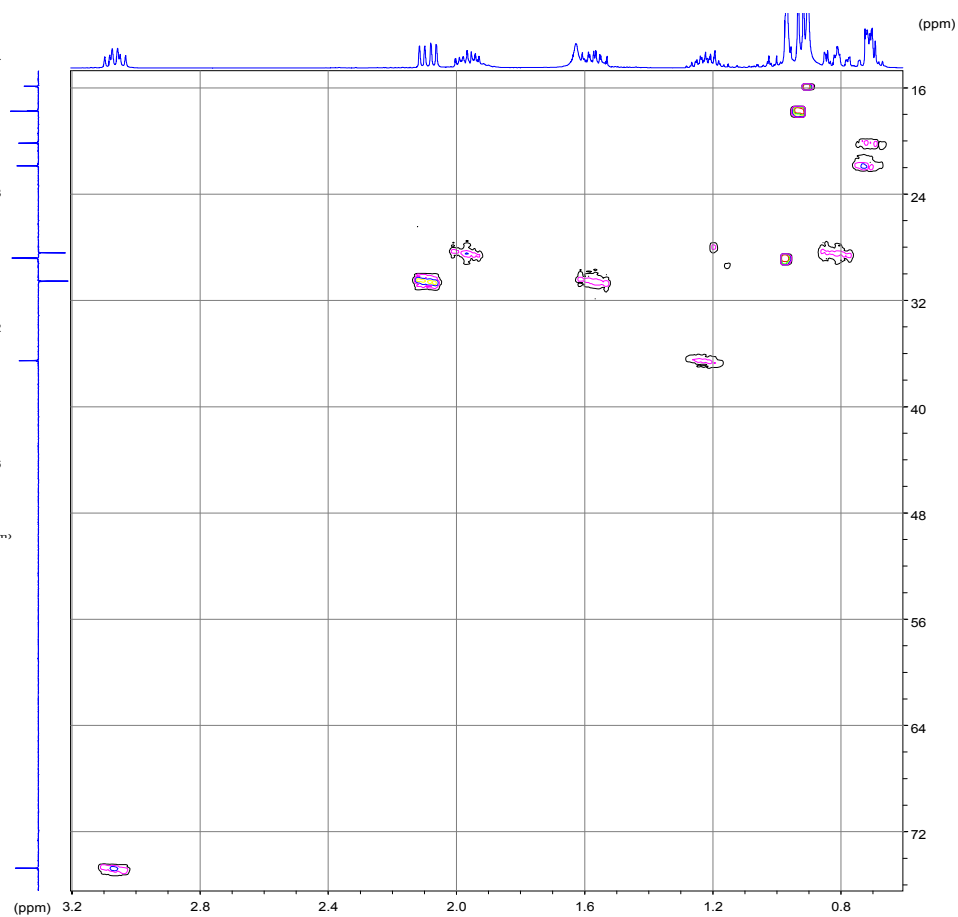
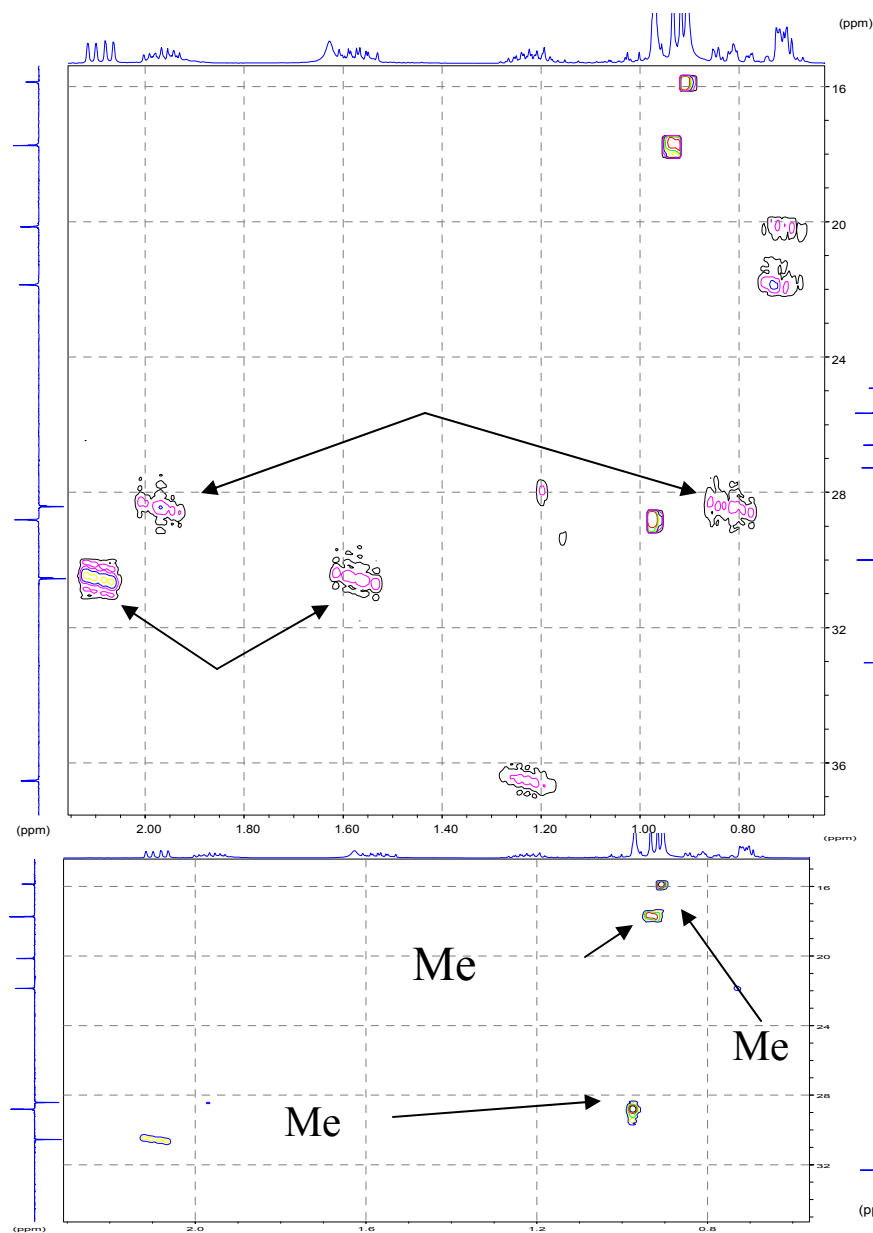
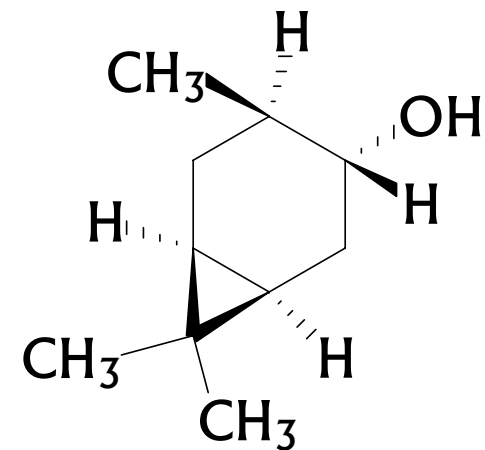
## 6A HH COSY of 1-Phenyl-1,4-butanediol



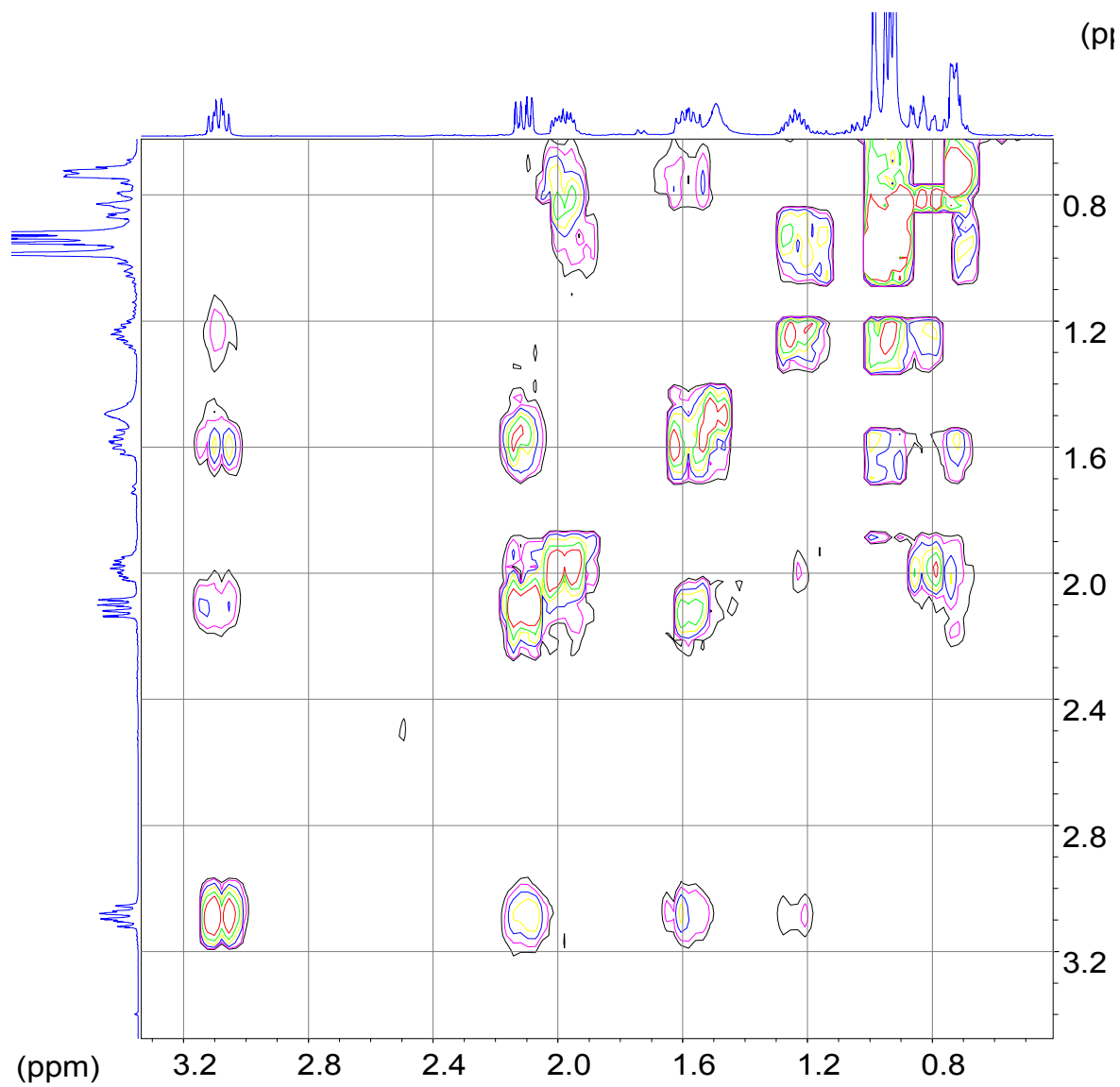
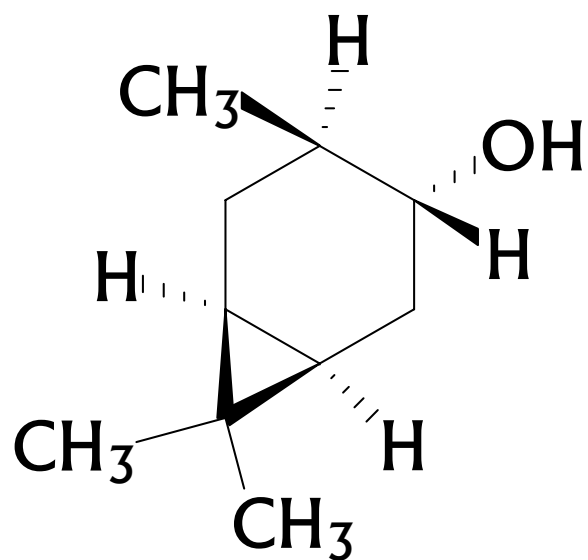
# 6B CH COSY $\gamma$ -Phenyl- $\gamma$ -butyrolactone



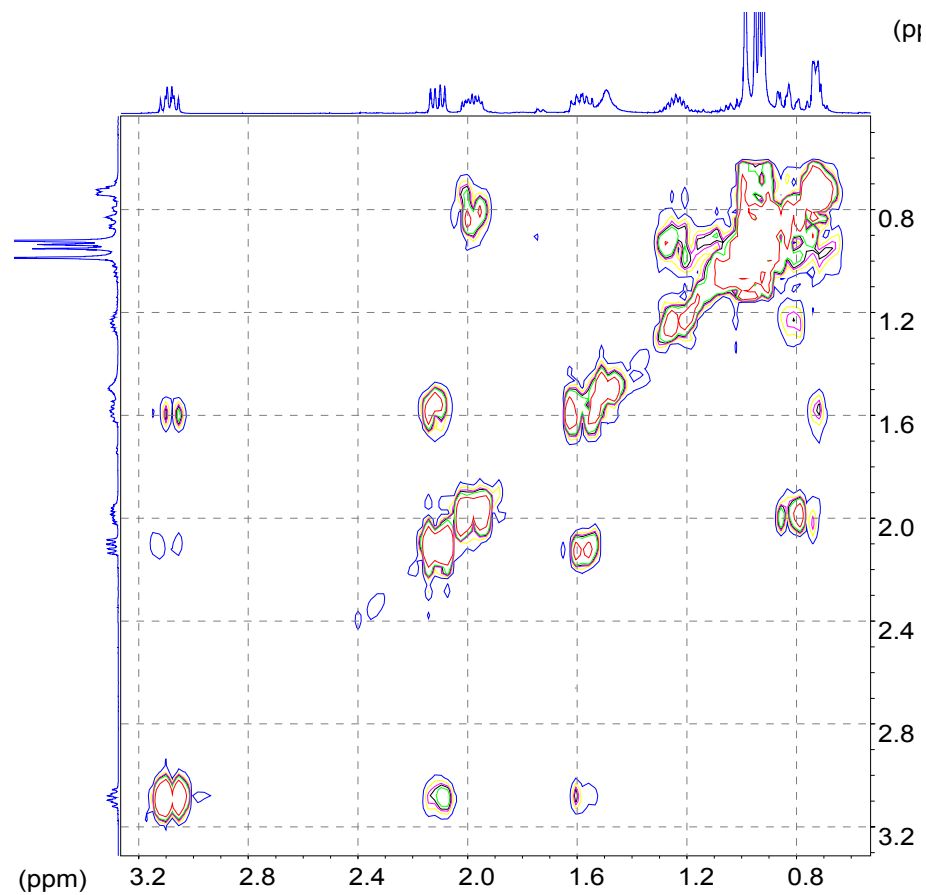
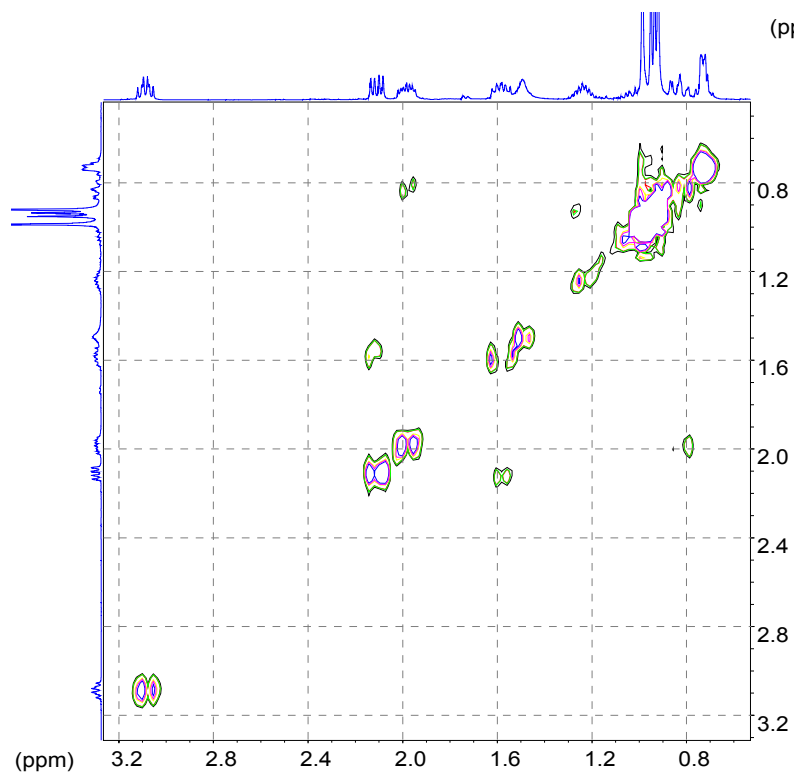
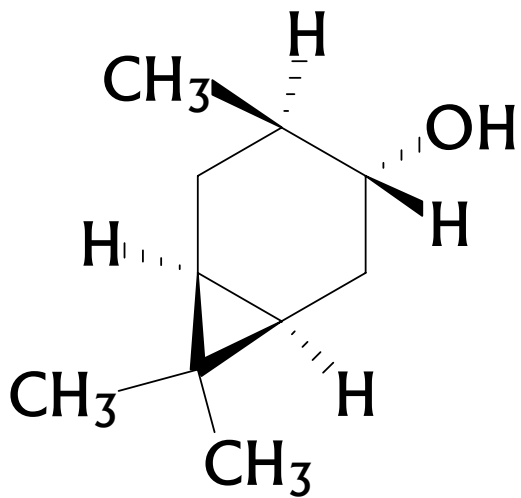
# 7 CH COSY *cis-caran-trans-4-ol*



# 7 HH COSY *cis*-caran-*trans*-4-ol



# 7 HH COSY *cis*-caran-*trans*-4-ol





# More complex NMR experiments to determine the configuration of the *cis-Caran-trans-ol*

## Long range CH COSY

used to find correlations to proton(s) other than those directly attached to a carbon  
used to establish links to hydroxy groups and 'quaternary' carbon peaks

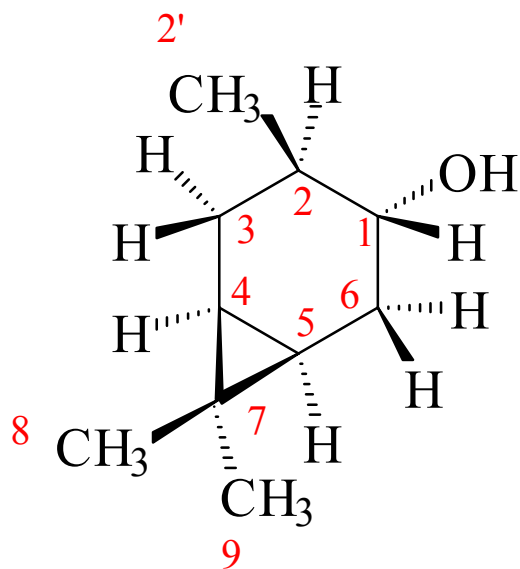
## Nuclear Overhauser Effect (NOE)

This will establish interactions of the spins through **SPACE**

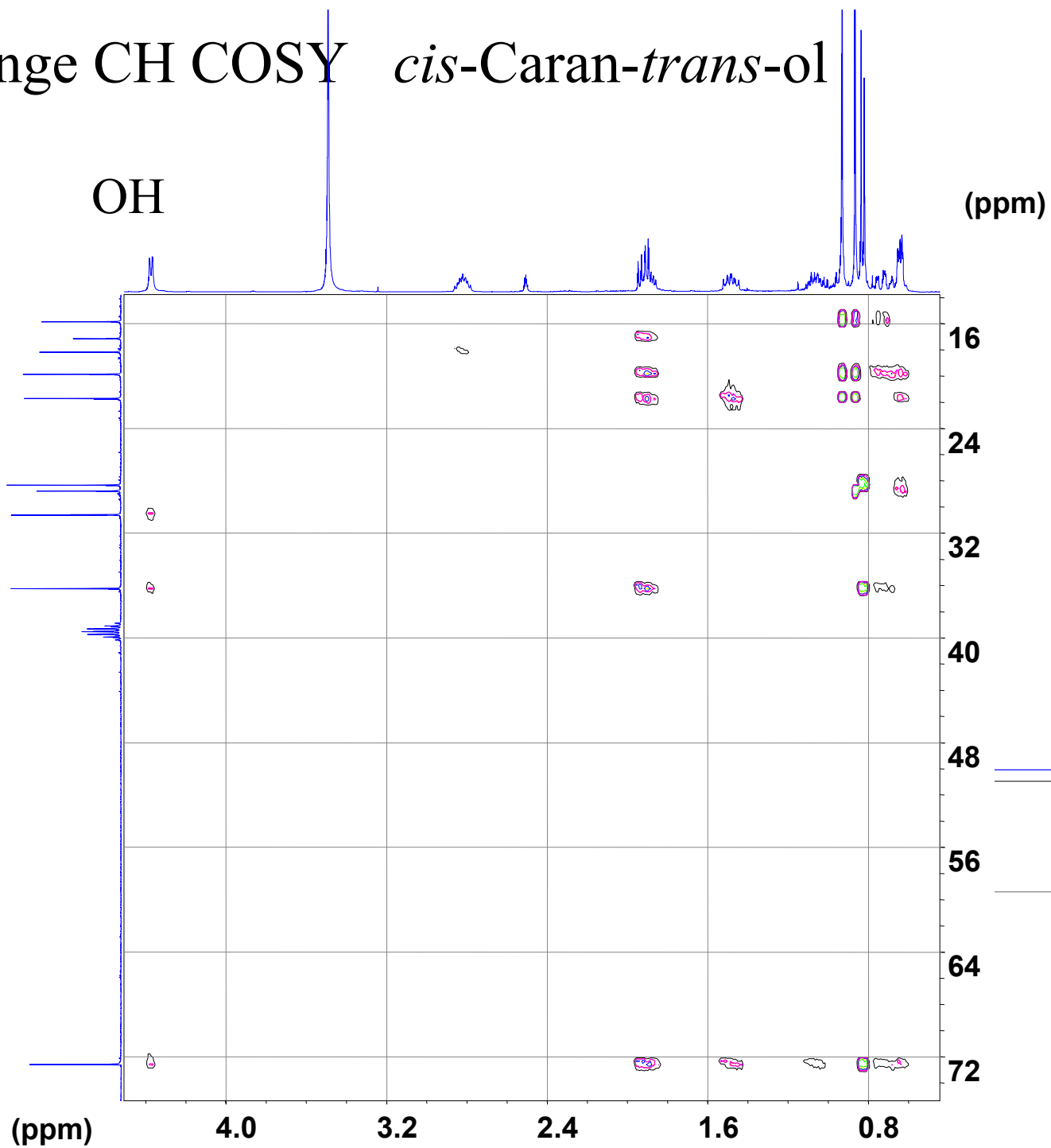
Difference NOE experiment – irradiate a specific proton and observe any changes  
(a normal spectrum is subtracted from the irradiated spectrum to give the DIFFERENCE spectrum)

expect to differentiate between the two *methyl peaks* on the *cyclopropyl* ring  
one should be lying in the same plane as the two *cis* protons on the ring

# 7 Long range CH COSY *cis*-Caran-*trans*-ol



Solvent DMSO-d<sub>6</sub>



## 7 Difference NOE NMR *cis*-Caran-*trans*-4-ol

